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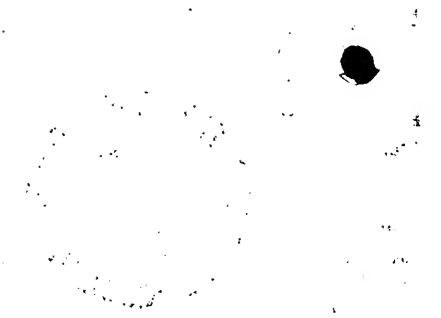
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*W. Evans*

Dated

*1st February 2002*

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145EP01 E659714-3 002898  
P01/7700 0.00-0122150.6

2. Patent application number

(The Patent Office will fill in this part)

13 SEP 2001

**0122150.6**

3. Full name, address and postcode of the or of each applicant (*underline all surnames*)

UNILEVER PLC  
UNILEVER HOUSE, BLACKFRIARS  
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Patents ADP number (*if you know it*)

00001628002

If the applicant is a corporate body, give the country/state of its incorporation

UNITED KINGDOM

4. Title of the invention

**EFFERVESCENT BEVERAGE PRODUCT**

5. Name of your agent (*if you have one*)

EVANS, Jacqueline Gail Victoria

"Address for Service" in the United Kingdom to which all correspondence should be sent (*including the postcode*)

PATENT DEPARTMENT, UNILEVER PLC  
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Patents ADP number (*if you know it*)

08094518001

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Country	Priority application number ( <i>if you know it</i> )	Date of filing ( <i>day / month / year</i> )

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Date: 13/09/01

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Name and daytime telephone number of person to contact in the United Kingdom

Isobel Howell, Tel 01234 22 2082

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## EFFERVESCENT BEVERAGE PRODUCT

5 This invention relates to a beverage product that comprises a dispenser and a pressurised beverage with a sparingly soluble effervescence inducing gas dissolved therein. The dispenser has a valve that can be opened to allow for the beverage to be discharged as an effervescent fluid.

10

Background and prior art

Carbonated beverages such as mineral water, soft drinks and iced teas are very popular with consumers. Carbon dioxide is readily soluble in water, it has a solubility of  $1.69 \text{ g kg}^{-1}$  in water at  $20^\circ \text{ C}$  and atmospheric pressure), and it is cheap, widely available and non-harmful. When carbon dioxide is mixed under pressure with a drinkable liquid in a container a substantial proportion of the gas remains dissolved in the drinkable liquid when the container is opened to the atmosphere. That enables the container to be resealed and stored for a short time if desired, with minimal deterioration to the gaseous liquid. These carbonated beverages are usually contained in aluminium cans or bottles made of glass or plastic.

The presence of a large amount of carbon dioxide in solution in a beverage produces a distinctive taste of carbonic acid. Some consumers find both the prickly sensation and the taste of carbonic acid to be unpleasant. It can also produce an uncomfortable bloating feeling.

Gases that are sparingly insoluble in water, such as oxygen, are not usually employed in a drinkable liquid in the same manner as more soluble gases such as carbon dioxide. Oxygen has a

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- 2 -

solubility of 0.043 g kg<sup>-1</sup> in water at 20° C and atmospheric pressure. As a result, when oxygen is added to a drinkable liquid under atmospheric conditions, or a container holding a pressurised oxygenated beverage is opened to the atmosphere, a substantial  
5 proportion of the oxygen contained in the beverage and packaging will rapidly escape from the beverage and be dispersed in the surrounding atmosphere.

Notwithstanding these difficulties, several attempts have been  
10 made to make a beverage that contains dissolved oxygen.

Japanese patent specification JP 64-27458 discloses a health drink product comprising a beverage with oxygen dissolved therein under pressure. However, since oxygen is very sparingly soluble in  
15 water and in the known health beverage nearly all of the oxygen packaged with the beverage escapes when the beverage container is opened. As a result, there is almost no effervescence on consumption.

20 United States patent specification US 5378480 describes a method for preparing an oxygenated beverage that is to be immediately ingested following preparation. The method involves measuring out in a container a quantity of powder that includes a foaming agent capable of trapping and temporarily holding oxygen gas, adding  
25 that powder to a predetermined quantity of liquid, evenly mixing the powder and liquid to produce a foamed beverage, and using a hand-held oxygen canister to evenly disperse oxygen throughout the beverage within the container via a dispensing tube. US 5378480 teaches that the beverage must be immediately ingested following  
30 preparation or else the oxygen content of the beverage will be rapidly lost.

Japanese patent specification JP 1168269 concerns a method of improving the flavour of a drink by generating bubbles in a drink  
35 that contains a large amount of oxygen. The drink is gasified by dispersing pressurised oxygen into a drink into which a high

concentration of oxygen has already been dissolved. The spray nozzle is operated to open the valve and release the water containing oxygen dissolved therein.

5 There therefore remains a need for a beverage product containing a sparingly soluble effervescence inducing gas that gives a smooth and silky mouthfeel when dispensed from a suitable dispenser that avoids at least some of the aforementioned disadvantages of the known products or at least provides the consumer with a useful  
10 alternative.

#### Statement of the invention

15 In broad terms the present invention relates to a beverage product comprising a dispenser and a beverage, the dispenser has a container for holding the beverage and a valve, the beverage has a sparingly soluble effervescence inducing gas dissolved therein; the beverage product being characterised in that the beverage is  
20 held under a gaseous pressure in the head space that is sufficient to cause the beverage to be discharged from the dispenser as an effervescent fluid when the valve is open.

The beverage is preferably held under a gaseous pressure in the  
25 headspace of at least 2.5 atmospheres gauge at 5 to 15° C.

The sparingly soluble effervescence inducing gas is preferably selected from the group consisting of oxygen, nitrogen, hydrogen, noble gases, gaseous hydrocarbons and mixtures thereof. If  
30 desired, a further gas which is more soluble in water than oxygen or nitrogen and which has a partial pressure in the headspace not exceeding 1 atmosphere absolute at 18° C, for example carbon dioxide, is dissolved in the beverage too.

35 The effervescent fluid contains very small bubbles of the sparingly soluble gas and does not give the prickly sensation

produced by effervescence of a more soluble gas such as carbon dioxide. The fluid provides the consumer with a smooth and silky sensation, which can enhance flavour delivery from a flavoured beverage. If the beverage is transparent, for example it is water  
5 or flavoured water, the many small bubbles of the sparingly soluble gas give the beverage a smoke-like appearance on consumption.

The beverage is preferably tea or coffee based, or water or a  
10 flavoured water.

Preferably the valve can be of a type which is known for use in pressurised dispensing containers (often known as aerosol cans). Such valves are readily available from several manufacturers.  
15 Known aerosol valves of this type have a valve member which is normally held in a sealing position to prevent egress of the contents of the pressurised container. In the sealing position the valve member is urged into contact with a sealing member, for example by a spring. The valve member is then moved to a  
20 dispensing position to allow the contents of the pressurised container to be dispensed. As the valve member is moved towards the dispensing position, the valve member and the sealing member are moved apart allowing the passage through the valve of the contents of the pressurised container under the pressure exerted  
25 by the propellant. The valve member may be moved away from the sealing member by a reciprocating movement or by a rocking movement. Though the above-described aerosol valves are preferred, any valve that is suitable for the purpose could be used in the present invention.

30 The dispenser preferably includes release means that are operable to open the valve to release the beverage from the container. The release means is preferably shaped and positioned to be engaged and operated by a consumer's mouth or teeth so that the  
35 effervescent fluid can be discharged directly into the consumer's



mouth.

The container preferably has a dip tube that extends from the valve into the interior of the container. It also has a closure member that normally closes the dip tube and is operable from the exterior of the container to open the tube.

The present invention can also be said to be a method for producing an effervescent beverage comprising holding a beverage with a sparingly soluble effervescence inducing gas dissolved therein in a container of a dispenser with a gaseous pressure in the headspace of the container that is sufficient to cause the beverage to be discharged from the dispenser as an effervescent fluid when the valve is open, and opening that valve to discharge the beverage as said effervescent fluid.

For the avoidance of doubt the word "comprising" is intended to mean including but not necessarily "consisting of" or "composed of". In other words the listed steps or options need not be exhaustive.

Except in the examples, or where otherwise explicitly indicated, all numbers in this description indicating amounts or concentrations of material ought to be understood as modified by the word "about".

#### Brief description of the drawings

The invention will now be further described, by way of example, with reference to the accompanying drawings, in which:

Figure 1 is a cross-sectional side view of a first preferred embodiment of beverage product of the present invention.

Figure 2 is a cross-sectional view to an enlarged scale of the valve of the beverage product shown in Figure 1.

5 Figure 3 is a cross-sectional side view of a second preferred embodiment of the beverage product of the present invention.

Figure 4 is a cross-sectional side view of a third preferred embodiment of the beverage product of the present invention.

10 Figure 5 is a cross-sectional side view of a fourth preferred embodiment of the beverage product of the present invention.

15 Figures 6a, 6b, 6c and 6d show internal and external views of a first preferred mouth operated actuator that can be attached to the neck of the dispenser.

Figures 7a, 7b, 7c and 7d show internal and external views of a second preferred mouth operated actuator that can be attached to the neck of the dispenser.

20 Figures 8a, 8b, 8c and 8d show internal and external views of a third preferred mouth operated actuator that can be attached to the neck of the dispenser.

25 Figures 9a and 9b show internal views of a fourth preferred mouth operated actuator that can be attached to the neck of the dispenser.

30 Figures 10a, 10b, 10c and 10d show internal and external views of a fifth preferred mouth operated actuator that can be attached to the neck of the dispenser.

35 Figures 11a and 11b show internal views of a sixth preferred mouth operated actuator that can be attached to the neck of the dispenser.

Figures 12a, 12b, 12c and 12d show internal and external views of a seventh preferred mouth operated actuator that can be attached to the neck of the dispenser.

- 5    Figures 13a, 13b, 13c and 13d show internal and external views of an eighth preferred mouth operated actuator that can be attached to the neck of the dispenser.

- 10   Figures 14a, 14b and 14c show cross-sectional views of a ninth preferred mouth operated actuator that can be attached to the neck of the dispenser.

- 15   Figure 15 is a cross-sectioned view of a fifth preferred embodiment of the beverage product of the present invention that includes a widget-type device.

Figures 16a and 16b are perspective views, from above and below respectively, of the widget-type device shown in Figure 15.

- 20   Figure 17 is a cross-sectional view of a sixth preferred embodiment of the beverage product of the present invention that has a gas capsule that is held within the container underneath the splayed end of the dip tube.

- 25   Figure 18 is a graph indicating acceptable pressures for the beverage product shown in Figure 3 (see Example 1).

- 30   Figure 19 is a view of an assembly used to test the effervescence of the second embodiment of the beverage product shown in Figure 3 (see Example 2).

Figure 20 is a graph illustrating the variation of bubble size with varying partial pressure of oxygen in the beverage product shown in Figure 3 (see Example 2).

Detailed description of the invention

The beverage product of the present invention comprises a dispenser and a pressurised beverage.

5

The dispenser has a container that is suitable for holding the beverage in a pressurised state and a valve that can be opened to allow for the beverage to be discharged from the dispenser as an effervescent fluid.

10

The beverage contains dissolved therein a sparingly soluble effervescence inducing gas. That gas can be oxygen, nitrogen, a noble gas or a mixture thereof. Oxygen is specially preferred as it is non-toxic, non-asphyxiating and safe at high temperatures.

15

If desired, a further gas which is more soluble in water than oxygen or nitrogen and which has a partial pressure in the headspace not exceeding 1 atmosphere absolute at 18° C, for example carbon dioxide, may be also dissolved in the beverage.

20

The sparingly soluble effervescence inducing gas should be dissolved in the beverage under a gaseous pressure in the headspace that is sufficient to cause the beverage to be discharged from the dispenser as an effervescent fluid when the valve is open. The present inventors have found that that

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pressure is preferably at least 2.5 atmospheres gauge at 5 to 15° C else the gas tends to be liberated from the beverage and not produce the desired effervescent fluid when the beverage is discharged from the dispenser.

30

The beverage is preferably tea or coffee based or water. It could include any suitable favouring substance or substances. The beverage can include milk, a whitener, a creamer, a flavour, a sweetener, a preservative or any number or combination thereof.

35

The beverage could also be fruit-based, vegetable-based, yoghurt-based, soya-based, alcoholic, a beer, a smoothie or a diet

product. If desired the beverage packaging can contain an oxygen adsorbing material such as a zeolite to minimise oxidation.

It is convenient for the valve of the dispenser to be an aerosol valve as they are readily available, however one could use any valve that is suitable for the purpose.

When the valve is opened the pressurised beverage discharges from the container of the dispenser as an effervescent fluid. This fluid contains very small bubbles of the sparingly soluble gas and does not give the prickly sensation or bloated feeling sometimes produced by effervescence of a more soluble gas such as carbon dioxide. The effervescent fluid provides the consumer with a smooth and silky sensation. It can even enhance the flavour delivery from a flavoured beverage. If the beverage is transparent, for example water or flavoured water, the effervescence of the sparingly soluble gas gives the beverage a fine, smoke-like appearance on consumption.

The dispenser preferably includes release means that is operable to open the valve to release the liquid from the container. The release means is preferably shaped and positioned to be engaged by a user's mouth or teeth so that the effervescent fluid will be delivered directly into the consumer's mouth. It could alternatively be adapted to be operated by the consumer's hand or finger. The release means can take a variety of forms but is preferably incorporated in an outlet portion of the dispenser. In a preferred embodiment of the dispenser the release means includes a button mounted in the nozzle. The button can be moved between a valve-closed position and a valve-open position by a twisting, sucking or biting action that is applied across the nozzle or by some other manner of manipulation by the consumer.

Various preferred embodiments of the dispenser of the beverage product of the present invention will now be described with reference to several preferred embodiments that are depicted in the accompanying drawings.

A first preferred embodiment of the beverage product of the present invention is shown in Figure 1. The beverage product comprises a dispenser 1 and a beverage 2. The dispenser 1 has a container 3 for holding the beverage. The container 3 is capable of withstanding an internal gauge pressure of at least 2.5 atmospheres. The beverage is water-based and has dissolved therein some of a sparingly soluble gas (oxygen, nitrogen, hydrogen, a gaseous hydrocarbon, one or more noble gases e.g. helium, neon and krypton, or a mixture thereof) that is contained within a headspace 4 above the beverage 2. The container 3 has a neck to which is fitted a normally closed valve 5. This is preferably a high-throughput aerosol valve. A dip tube 6 extends downwardly from the valve 5, through the headspace 4 and into the beverage 2. When the valve 5 is opened and the dispenser 1 is upright the pressurised beverage 2 will be urged up the dip tube 6 and expelled from the dispenser 1 through the valve 5.

In the first preferred embodiment the dip tube 6 is rigid however it could alternatively be made of a flexible material. When the dip tube 6 is flexible it can be advantageous to attach a weight to the end of the tube 6 so that it remains immersed in the beverage as the dispenser is tilted and the volume of the beverage drops.

The valve 5 of the first preferred embodiment is shown schematically in greater detail in Figure 2. The valve 5 has a valve housing 7, the lower end of which is attached to the upper end of the dip tube 6 as shown in Figure 1. The upper end of the housing 7 has a gasket 8 forming a valve seating that co-operates with a movable valve member 9. That valve member 9 is constituted by a stem 10 that is biased by a spring 11 to a closed position in which the stem 10 engages the gasket 8. A further gasket 12 and an annular ring 13 seal the valve 5 to the neck of the container

3.

To dispense the beverage 2, the user tilts or depresses the stem 10 (or another element attached to the stem 10) so as to overcome the bias of the spring 11 and allow the beverage to pass between the stem 10 and the gasket 8. The beverage 2 may be dispensed  
5 into another container for drinking but is preferably dispensed directly into the consumer's mouth.

A second preferred embodiment of the beverage product of the present invention is shown in Figure 3. The beverage product  
10 shown in Figure 3 is similar to that shown in Figure 1 except it does not have a dip tube 6. That means the dispenser must be inverted or at least significantly tilted before a consumer can drink from the dispenser.

15 In order to manufacture the first and second embodiments of the beverage product shown in Figures 1 and 3, the beverage 2 is placed in the container 3, which can then be flushed with the effervescence-inducing gas. The container 3 is then sealed with the valve 5, and more of the effervescence-inducing gas is added  
20 through the valve 5 until the pressure in the headspace 4 has reached its required value, which is preferably more than 2.5 atmospheres gauge. When the valve 5 is then opened for operation the pressure in the headspace 4 expels the beverage through the valve 5 as an effervescent fluid. Because of this, unless the  
25 pressure is very high, a substantial amount of headspace should initially be present. Preferably, the headspace 5 occupies 10 to 80%, preferably 20 to 50%, of the capacity of the container 2.

A third preferred embodiment of the beverage product of the present invention is shown in Figure 4. In this embodiment the  
30 dip tube 6 of the first preferred embodiment is replaced by a flexible bag 14 that has an upper neck that is sealed to the housing 7 of the valve 5. The bag 14 holds the beverage 2 under pressure from the sparingly soluble gas or gas mixture in the  
35 headspace 4 and also a second gas space 15. The headspace 4 can be considerably smaller than the headspace 4 of the beverage

products shown in Figures 1 and 3. The second gas space 15 exists between the bag 14 and the container 3. This second gas space 15 is filled with pressurised gas which may be the same gas (or a gaseous mixture) as that which fills in the headspace 4 or may be different. This construction allows the beverage 2 to be dispensed regardless of the orientation of the container 3.

In order to manufacture the third embodiment shown in Figure 4, the sparingly soluble gas is placed in the headspace 4, which may be very small, preferably less than 20% of the capacity of the bag 14. The space 15 between the bag 14 and the container 3 is pressurised after the valve 5 has been sealed to the container. The expansion of gas in this space provides the driving force to expel the beverage 2 from the container 3 of the dispenser 1 through the valve 5 and into another container for drinking or dispensing directly into the consumer's mouth.

In each of the aforementioned embodiments, the pressure in the headspace 4 exceeds 2.5 atmospheres gauge, which causes the sparingly soluble gas (or mixture of sparingly soluble gases) to effervesce on discharge into the consumer's mouth or into water under test conditions. This effect is demonstrated in Examples 1 and 2.

A fourth preferred embodiment of the beverage product of the present invention is shown in Figure 5. It is very similar to the first embodiment but differs in that the container has a curvaceous shape to more ergonomically fit the hand of a consumer, and more significantly in that an aperture 16 is provided in the wall of the dip tube 6 or the valve housing 7 that is preferably close to where the dip tube 6 is attached to the valve 5. While in this embodiment the aperture 16 takes the form of a single aperture, one could alternatively have multiple apertures or use a mesh or some other form of porous material.



In use, with the dispenser 1 in the upright position, a consumer opens the valve 5 by applying downward or sideways force on valve member 9. If a non-aerosol-type valve were used the consumer might apply a torque to open it instead. On opening of the valve 5, gas is entrained through the aperture 16 in dip tube 6 into the stream of pressurised beverage 2 that flows through the dip tube 6 resulting in the expulsion of a two-phase mixture of liquid and gas that constitutes an effervescent fluid. When force is removed from the valve member 9 the valve 5 closes under the action of the spring 11, terminating the flow of the effervescent fluid from the container 3. The valve 5 can be operated in this manner as often as desired until the supply of beverage and gas has been exhausted. Unlike the first preferred embodiment, the fourth preferred embodiment functions in any orientation, but it tends to work better when held upright.

A nozzle is preferably attached to and forms part of the dispenser 1 of the beverage product of the present invention and includes means for moving the valve 5 from its closed position to its open position. This actuation is preferably reversible and more especially operable using the consumer's mouth and/or teeth.

Figures 6a through to 14c depict a variety of mouth operated actuators that will now be described in detail.

A first preferred embodiment of a mouth operated actuator 20 is shown in Figures 6a, 6b, 6c and 6d. Referring to Figure 6a, which is a cross-sectional view, the actuator 20 is secured to the container 3 and/or the valve 5 of a dispenser 1 and covers the stem 10 of an aerosol valve 5. The actuator 20 has a housing 21 and a stem extension 24 that are pivotally attached to the dispenser at pivot point 22.

The upper part of the housing 21 forms a tapering nozzle section 23 that has a central passage constituted by a stem extension 24 that leads from the valve stem 10 to a nozzle outlet 25 at the

upper end of the nozzle. The housing 21 and the nozzle section 23 are shaped so that when it is inserted into the mouth of a consumer, the consumer's lips engage the nozzle to provide a seal.

- 5 The actuator has a button 26 that is attached to the stem extension 24. The button 26 protrudes from the tapering wall of the housing 21 through a circular button aperture 27. The button 26 is movable between a normal inoperative position shown in Figure 6a and a depressed operative position shown in Figure 6b.
- 10 In the position of the button illustrated in Figure 6a, the valve is closed so no effervescent fluid can escape from the container 3. In the position of the button illustrated in Figure 6b, the button 26 has displaced the valve stem 10, pressing it down vertically by means of the stem extension 24 rotating about the
- 15 pivot or hinge point 22. Thus the valve stem 10 is linearly displaceable in this embodiment. The area of the nozzle opposite to the button 26 is formed with a textured surface 28a, e.g. by the integral moulding of recesses or projections or by the application of a piece of non-slip material. A similar textured
- 20 surface may be provided on the surface 28b of the button 26.

Figures 6c and 6d show front and rear exterior views of the mouth operated actuator shown in Figures 6a and 6b.

- 25 In use a consumer grasps the outer case of the dispenser 1 and inserts the nozzle section 23 of the mouth operated actuator into his or her mouth. To open the valve and thus dispense the effervescent fluid directly into the mouth, the consumer applies a biting action on the textured surface 28b on the button 26 and the
- 30 textured area 28a on the area opposite the button 26. That action depresses the button 26 in the direction of movement illustrated by the solid arrow in Figures 6a and 6b. The textured surfaces 28a and 28b prevent the consumer's teeth from slipping. The button 26 reverts to its normal extended position (and the valve
- 35 to its closed position) when the compressive force applied across the nozzle as a result of the biting action is removed.

A second preferred embodiment of a mouth operated actuator is shown in Figures 7a, 7b, 7c and 7d. In this embodiment the valve is tilt operable. The button 26 of the actuator 20 shown in  
5 Figures 7a and 7b is not attached to the stem extension 24 but is attached by hinge point 29 to the housing 21. When the button 26 is engaged by the consumer's teeth it pivots about the hinge point 29 and this displaces the valve stem 10, thereby opening the valve so as to dispense the effervescent fluid directly into the  
10 consumer's mouth. The stem extension 24 may be made of a flexible material, so that it can flex and thereby maintain a good seal with the valve stem 10 during operation.

The external appearance of the second preferred embodiment of the  
15 mouth operated actuator of Figures 7a and 7b is shown in Figures 7c and 7d, as front and rear views respectively.

A third preferred embodiment of a mouth operated actuator is shown in Figures 8a, 8b, 8c and 8d. In this embodiment the actuator 20  
20 has two depressible buttons 26 on opposite sides of the housing 21, each button being like the button 26 of the second preferred embodiment of the actuator shown in Figures 7a and 7b. When the two buttons are depressed (as seen in Figure 7b), they pivot about hinge points 29 and engage a downwardly diverging frusto-conical  
25 shape at the end of the stem extension 10, so as to depress the stem 10 and open the valve, thereby dispensing the effervescent fluid. Thus, the stem 10 is linearly displaceable in this embodiment.

Figures 8c and 8d show, in front and rear views respectively, the  
30 external appearance of the nozzle of the dispenser of Figures 8a and 8b.

A fourth preferred embodiment of a mouth operated actuator is  
35 shown in Figures 9a and 9b. In this embodiment the button 26 is mounted on the stem extension 24 and when it is depressed (as seen

in Figure 9) it tilts the stem extension 24 and opens the valve thereby dispensing the effervescent fluid. The external appearance is the same as that of the first preferred embodiment of the actuator shown in Figures 6c and 6d.

5

A fifth preferred embodiment of a mouth operated actuator is shown in Figures 10a, 10b, 10c and 10d. In this embodiment the button 26 can be lower down on the housing 21, when depressed, the stem extension 24 pivots about a fulcrum point 30 near the top of the housing 21.

10

Figures 10c and 10d show the external appearance of the actuator, in front and rear views respectively.

15

Alternatively, the top of the stem extension can be firmly fixed to the top of the housing 21, and the stem extension 24 can be made of a flexible material. In such a variation, the button 26 attached to the stem extension 24 is typically higher up, and when the button is depressed, it causes the stem extension 24 to flex and at the same time the valve stem 10 to tilt, thereby opening the valve 5. The flexibility of the stem extension 24 allows a good seal to be maintained between the valve stem 10 and the stem extension 24 during operation.

20

25

A sixth preferred embodiment of a mouth operated actuator is shown in Figures 11a and 11b. In this embodiment the actuator has a flexible stem extension 24. Depressing button 26 causes the stem extension 24 to flex, thereby tilting the valve stem 10 and opening the valve. The button 26 is hinged to the housing 21 at hinge point 29.

30

The external appearance of the sixth preferred embodiment of actuator is the same as that of the second preferred embodiment of actuator as shown in Figures 7c and 7d.

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A seventh preferred embodiment of a mouth operated actuator is shown in Figures 12a, 12b, 12c and 12d. In this embodiment the consumer grips the housing 21 with his or her teeth and applies a downward force in the direction of the solid arrows shown in  
5 Figures 12a and 12b. The housing 21 has a concertina formation 31 that allows the upper part of the housing 21 to be moved towards the container. The downward force applied to the housing 21 is transferred to the valve stem 10 so as to open the valve, thereby dispensing the effervescent fluid.

10 Figures 12c and 12d illustrate the external appearance of the seventh preferred embodiment of the actuator in front and rear views respectively and show the aforementioned concertina formation 31 of the housing 21.

15 An eighth preferred embodiment of a mouth operated actuator is shown in Figures 13a, 13b, 13c and 13d. In this embodiment the nozzle section 23 is a separate part of the actuator 20 that is threaded onto the housing 21 to provide for relative rotation of  
20 the nozzle section 23 and the remainder of the dispenser 1. In use the consumer grips the nozzle section 23 between his or her teeth and then grasps the housing 21 (where it surrounds the container) so as to rotate the container 3 relative to the nozzle section 23. This rotation (indicated by the solid arrows in  
25 Figure 13b) moves the valve stem 10 to open the valve and thereby dispense the effervescent fluid directly into the consumer's mouth.

30 Figures 13c and 13d illustrate the external appearance of the eighth preferred embodiment of actuator, in front and rear views respectively.

A ninth preferred embodiment of a mouth operated actuator is shown in Figures 14a, 14b and 14c. In this embodiment one side of the  
35 actuator has a concave recess 32 that is designed to contact the lower lip of a consumer. But more importantly the actuator 20 has

a slidable plug 33 that is formed around the stem extension 24 and prevents the valve from being opened when the dispenser is inverted.

5 The actuator is similar to that of the sixth preferred embodiment shown in Figures 11a and 11b in that the stem extension 24 is flexible and the valve 5 is opened by applying pressure on the button 26, which causes the stem extension 24 to flex and thereby tilt the valve stem 10. The slidable plug 33 has no effect when  
10 the dispenser is used in a substantially upright position (see Figures 14a and 14b). However, when the dispenser is inverted (see Figure 14c) the slidable plug 33 slides down the stem extension 24 towards the nozzle outlet 25 and a wedge 34 that forms part of the plug 33 provides sufficient resistance to  
15 prevent the consumer from flexing the stem extension 24 when he or she tries to depress the button 26. A stopper 35 is provided on the exterior surface of the stem extension 24 adjacent to the nozzle 23 to restrict the movement of the plug 33 when the dispenser is inverted.

20 In all embodiments of the mouth operated actuator, the delivery of liquid is directly into the consumer's mouth, under the control of the consumer's bite. The consumer's fingers do not need to touch any part of the nozzle, thereby reducing the chance of the nozzle  
25 becoming contaminated. If desired a protective and replaceable cap (not shown) can cover the nozzle.

If desired the beverage product of the present invention can include a widget that contains the sparingly soluble gas or gas  
30 mixture or a flavour concentrate that is released into the container when the dispenser is being used.

"Widget" has become a term of art, at least in the brewing industry, to describe a device that releases pressurised gas into  
35 a can of beer when the can is opened thereby producing a creamy head typical of draught beer. The first widget or in-can-draught

system is described in United Kingdom patent specification GB 2183592 A (Arthur Guinness Son and Company (Dublin) Limited). Various other versions have since become commercially available.

- 5 A fifth preferred embodiment of the beverage product of the present invention includes a widget-type device, herein just called a "widget". The embodiment is shown in Figure 15 and the widget is shown in more detail in Figures 16a and 16b.
- 10 The fifth embodiment of the beverage product shown in Figure 15 is much like the first embodiment shown in Figure 1 in that it comprises a dispenser 1 for containing a beverage 2 in a container 3. The dispenser has a valve 5 (not shown) with a dip tube 6. In the fifth embodiment a toroidal widget 36 is suspended above the
- 15 base of the container 3 around the inlet of the dip tube 6. The widget is shown in a perspective view from above in Figure 16a and in a perspective view from below in 16b. Perforations 37 are formed in the downward facing surface of the widget 36, as seen in Figure 16b. Naturally the perforations could take other forms; a
- 20 single perforation could be sufficient or the widget could include a mesh or some other porous material. In the case where the widget 36 shown is filled with just gas the perforations 37 can face in any orientation.
- 25 The widget 36 contains the sparingly soluble gas or gas mixture under pressure and remains there while the contents of the container 3 is pressurised. In use when a consumer opens the valve for the first time the pressure within the container drops as pressurised gas within the headspace escapes. The sparingly
- 30 soluble gas or gas mixture escapes from the widget 36 through the perforations 37. With the sparingly soluble gas or gas mixture free within the container the effervescent fluid will be created as previously described, which can be poured into a drinking vessel or directed straight into the consumer's mouth.
- 35

A sixth preferred embodiment of the beverage product of the present invention that also includes a widget is shown in a cross-section in Figure 17. In this embodiment the widget 36 is substantially spherical in shape and includes at least one perforation 37 that is located to discharge its contents upwards into a dip tube 6. The dip tube 6 has a splayed or flared inlet 38 to catch the discharging contents of the widget 36. The dip tube 6 optionally has an aperture 16 much like the fourth preferred embodiment of the beverage product.

If desired the widget, such as the widget shown in Figure 16, a can contain a concentrated flavour together with the sparingly soluble gas or gas mixture. In this case the flavour is expelled from the widget close to the end of the dip tube and so is taken together with the beverage. This arrangement can be used to keep labile flavours separate from the gas in the headspace above the beverage

#### EXAMPLES

The beverage product of the present invention will now be described with reference to the following examples.

#### EXAMPLE 1

##### Determination of filling pressure for effective effervescence

The following experiment was conducted to determine the filling pressure that is required to produce the desired effervescence.

Gas samples containing oxygen and nitrogen in various proportions (molar fraction of O<sub>2</sub> of 0.0, 0.25, 0.5, 0.75, 1.0) were prepared and used to fill a beverage dispenser as shown in Figure 3 using a variety of filling pressures. The dispensers were tested to see whether they produced a fine smoke-like effervescent fluid.



The samples were prepared and tested in duplicate to check reproducibility. All temperatures given were measured using a thermocouple probe.

5 For each sample, the following protocol was observed:

- 10 (1) 300 ml of tap water at  $10 \pm 0.5^\circ \text{C}$  was poured into a strong polyethylene terephthalate ("PET") bottle of 28 g mass and 520 ml brimful capacity. The bottle was then left to equilibrate in a temperature-controlled bath also at  $10 \pm 0.5^\circ \text{C}$ .
- 15 (2) The proportion of oxygen gas in the oxygen-nitrogen mixture had been decided on in advance, and whichever gas was to be at a higher partial pressure in the mixture is called "gas 1". The other gas will be referred to as "gas 2".
- 20 (3) The headspace above the water in the bottle was flushed with gas 1 for 5 to 10 seconds, and then the aerosol valve was placed quickly over the neck of the bottle, thus sealing the bottle from any loss of gas 1 or influx of air. The aerosol valve was then held firmly in place by means of a screw cap with a small hole cut in the centre in order to allow the stem of the valve to be accessible.
- 25 (4) More of gas 1 was then added from the regulated supply of gas 1 through the stem of the aerosol valve until a pressure "p1" was reached. The bottle was then shaken to ensure that the gas inside was at the same temperature as the liquid, and the bottle was topped up to the pressure p1 with more of gas 1. This was necessary because the temperature of the gas when first injected into the bottle is somewhat higher than  $10^\circ \text{C}$ .
- 30 (5) Gas 2 was then added through the stem of the valve from a regulated supply of gas 2 until a pressure "p2" was reached.

Again the bottle was shaken, and then topped up with more of gas 2 to reach the pressure  $p_2$  again. Both  $p_1$  and  $p_2$  are measured in atmospheres gauge from the pressure gauge on the regulated supplies of the two gases.

5

- (6) The sealed bottle was then placed in a temperature controlled bath at  $10 \pm 0.5^\circ \text{C}$  for 5 minutes to equilibrate, the sample being shaken every 30 seconds to ensure good contact between the gas mixture and the water in the bottle.

10

- (7) The bottle was then discharged under water that was at a temperature of  $21 \pm 1^\circ \text{C}$ . The stream of discharged fluid was observed as it left the bottle against a black plastic sheet, and it was noted whether or not there was a visible, fine, smoke-like effervescence.

15

The results of this experiment are summarised in Figure 18 as a plot of the molar fraction of oxygen present in the mixture (horizontal axis) against the gauge pressure  $p_2$  (vertical axis).

20

Each error bar in Figure 18 runs from a lower value of  $p_2$ , which is the highest pressure at which the effervescence is definitely not seen, to a higher value of  $p_2$ , which is the lowest pressure  $p_2$  at which effervescence is definitely seen. The true critical pressure thus lies somewhere on the error bar. If there is a pressure at which there is some doubt as to whether effervescence can be seen or not, this is marked as an "x" on the error bar. These "x" marks thus represent the best estimate of the position of the critical filling pressure.

30

The graph has two results drawn with a lighter weight of line and marked 40 and 41. These results are shown slightly displaced in the horizontal direction, for the sake of clarity. Thus the true molar fraction of oxygen in the mixture for result 40 is 0.0, and for result 41 it is 1.0. The results labelled 40 and 41 are for a tilt operated aerosol type valve, while all the other results are

35

for a push-down operated aerosol type valve. These results show that there is no significant difference between the types of valves used.

- 5 The results show that fine smoke-like effervescence is produced from a beverage that has an initial pressure in the headspace of 2.5 atmospheres or more, for all mixtures of oxygen and nitrogen.

## EXAMPLE 2

### 10 Effervescence obtained using gas mixtures

The following experiments were conducted to determine whether the desired effervescence is obtained using a gas mixture that contains a gas that is sparingly soluble in water as well as a gas  
15 that is soluble in water.

Twelve experiments were conducted in accordance with the following protocol. The experimental set-up is shown in Figure 19 and involves using the second preferred embodiment of the beverage  
20 product shown in Figure 3. All temperatures given were measured using a thermocouple probe.

- (1) 300 ml of tap water at  $10 \pm 0.5^\circ \text{C}$  was poured into a strong polyethylene terephthalate ("PET") container (labelled 3 in  
25 Figure 19) of 28 g mass and 520 ml brimful capacity. The container 3 was then left to equilibrate in a temperature-controlled bath also at  $10 \pm 0.5^\circ \text{C}$ .
- (2) Twelve experiments were performed, referred to in the  
30 following by letters "A" to "L" inclusive. These experiments differed in the way gas was added to the container 3. In all experiments the valve 5 of the dispenser 1 shown in Figure 19 was secured to the container 3 by means of a screw cap. The screw cap had a hole through

which the valve stem 10 could extend but gave a gas-tight seal.

5 In Experiment A oxygen was added through the valve stem 10 until the pressure inside the headspace 4 was 4 atmospheres gauge. The container 3 was shaken, and then topped up with oxygen in the same way until the pressure inside was once more 4 atmospheres gauge.

10 In Experiment B the headspace 4 of the container 3 was flushed with carbon dioxide from a regulated supply of the gas, for 5 seconds, to clear the headspace 4 of air. The container 3 was then sealed and CO<sub>2</sub> was added at 4  
15 atmospheres from the regulated supply through the valve stem 10. While the gas was added, the container 3 was constantly shaken for 20 seconds, so that the carbon dioxide was close to equilibrium with the water 40.

20 In Experiment C the procedure of Experiment A was followed except before the container 3 was sealed the headspace 4 was flushed with oxygen for 5 seconds to remove the air from the headspace 5.

25 In Experiments D and J the container 3 was flushed with oxygen and sealed as in Experiment C. Oxygen gas was then added through the valve stem 10 up to a pressure of 3  
30 atmospheres gauge in Experiment D and up to 3.5 atmospheres gauge in Experiment J. The container 3 was then shaken and topped up to 3 atmospheres gauge in Experiment D and 3.5  
35 atmospheres gauge in Experiment J with oxygen. In both cases carbon dioxide was then added through the valve stem 10 with continuous shaking for 20 seconds up to a final pressure of 4 atmospheres gauge.

In Experiment E the headspace was flushed for 20 seconds with CO<sub>2</sub>, while the container 3 was shaken, so that the CO<sub>2</sub>

was close to equilibrium with the beverage 2 (i.e. water). The container 3 was then sealed and oxygen added through the valve stem 10 to a pressure of 4 atmospheres gauge. The container 3 was shaken, and topped up to 4 atmospheres gauge with oxygen.

In Experiments F, H and I the container 3 was flushed for 5 seconds with CO<sub>2</sub> then sealed. Carbon dioxide was then added through the valve stem 10, with shaking, for 20 seconds. The final pressure of CO<sub>2</sub> was chosen to be 1, 2 and 3 atmospheres gauge for the Experiments F, H and I respectively. The containers were then topped up with oxygen to 4 atmospheres gauge in all cases, shaken and topped up to 4 atmospheres gauge again with oxygen.

In Experiments G and K (which are identical) the container 3 was flushed with CO<sub>2</sub> for 5 seconds to expel any air present, but was not shaken. The container 3 was then sealed and then oxygen was added through the valve stem 10 up to a pressure of 4 atmospheres gauge. The container 3 was then shaken, and topped up with oxygen to 4 atmospheres gauge.

In Experiment L the procedure of Experiment C was followed except nitrogen was used in place of oxygen at all points.

- (3) At this stage of the procedure, the containers were all filled to 4 atmospheres gauge pressure, but with different mixtures of gases.
- (4) In each experiment, the container 3 was allowed to equilibrate in a temperature-controlled bath at  $18 \pm 0.5^\circ \text{C}$ , for 5 minutes, with shaking every 30 seconds (as in Experiment 1).
- (5) In each case the container 3 was inverted, and a small fraction of its contents was discharged through a first

flexible rubber tube 45, which made a good seal with the valve stem 10. The rubber tube 45 was connected to a glass pipe 46, which was fixed horizontally. Most of the length of the glass pipe consisted of a flat-sided pipe of height 9 mm (vertically), width 5 mm and length 15 cm. The other end of the glass pipe was connected to a second flexible rubber tube 47, which emptied into a beaker 48. The function of the flat-sided pipe was to enable the liquid discharged from the container 3 to be clearly seen and imaged.

Whilst the liquid was being discharged, a video camera was used to record the appearance of the discharged fluid a distance 15 cm from valve 3. This viewing region is labelled 49 in Figure 19. A grid with lines separated by 1 mm was placed behind the glass pipe to allow distances to be measured accurately from the video images.

(6) After a short discharge through the valve, effervescence was observed in the viewing region 49 of the glass pipe 46 using the video camera. This consisted of bubbles that rose in the glass pipe 46. When they reached the top of the glass pipe, the bubbles coalesced into larger bubbles. However, while they were rising in the now stationary fluid, their rise velocities could be measured by counting the number of video frames required for the bubbles to rise 1 mm. These velocities were converted into bubble sizes using Stokes' law for the rising of a sphere in a liquid of known density and viscosity (in our case, water at 18° C).

The results of these twelve experiments are summarised in Table 1 below. The table refers to "Single" and "Multiple" rise times. The "single" rise time is the rise velocity of individually imaged bubbles. Whereas the "Multiple" rise time is the rise velocity of the surface of the cloud of effervescence.

TABLE 1

Results of Experiment 2

Expt	Partial pressure of oxygen	Type of rise Velocity measured	No. of readings taken	Bubble diameter
A	4.2	Single	9	48 ± 17
A	4.2	Multiple	6	39 ± 3
B	0.0	Single	9	148 ± 30
C	5.0	Single	14	49 ± 10
C	5.0	Multiple	3	55 ± 12
D	4.0	Single	33	80 ± 8
D	4.0	Multiple	8	75 ± 5
E	4.0	Single	25	61 ± 8
E	4.0	Multiple	11	61 ± 7
F	3.0	Single	37	101 ± 13
F	3.0	Multiple	7	88 ± 7
G	4.5	Single	29	51 ± 11
G	4.5	Multiple	9	48 ± 5
H	2.0	Single	28	123 ± 18
I	1.0	Single	12	181 ± 31
J	4.5	Single	52	72 ± 8
J	4.5	Multiple	11	66 ± 5
K	4.5	Single	45	54 ± 11
K	4.5	Multiple	10	49 ± 5
L	0.0	Single	11	31 ± 2
L	0.0	Multiple	12	28 ± 3

5

The results for the mixtures of oxygen and carbon dioxide (that is to say, for experiments B, C, D, E, F, G, H, J, and K) are plotted giving the graph shown in Figure 20. In Figure 20, the results from "Single" bubble rise times are plotted as filled circles, and the results from "Multiple" bubble rise times are plotted as open circles.

15 It is clear from these results that the bubble size of the effervescence increases from the value for pure oxygen if a large fraction of carbon dioxide is present in the gas mixture.

Experiments A and L show that fine effervescence (like that produced by pure oxygen) is also produced for another sparingly soluble gas (nitrogen), and also a mixture of nitrogen and oxygen.

- 5 Experiments B to K inclusive show that if the partial pressure of the soluble gas (carbon dioxide) does not exceed 1 atmosphere absolute (that is to say, if the partial pressure of the sparingly soluble gas (oxygen in this case) is greater than or equal to the total absolute pressure above the beverage, minus one atmosphere),  
10 then the effervescence size does not differ significantly from the value for pure oxygen.

- In other words, the fine smoke-like effervescence produced by mixtures of nearly insoluble gases that gives a smooth (as opposed  
15 to prickly) sensation to the consumed beverage, is preserved in the presence of admixtures of a soluble gas. That is provided that the partial pressure (absolute) of the soluble gas above the beverage in the container does not exceed atmospheric pressure. That is to say, provided the soluble gas is not overly  
20 supersaturated at the serving pressure of 1 atmosphere absolute pressure.



## CLAIMS

1. A beverage product comprising a dispenser and a beverage,  
the dispenser has a container for holding the beverage and a  
5 valve, the beverage has a sparingly soluble effervescence  
inducing gas dissolved therein, the beverage product being  
characterised in that the beverage is held under a gaseous  
pressure in the head space that is sufficient to cause the  
beverage to be discharged from the dispenser as an  
10 effervescent fluid when the valve is open.
2. A beverage product according to claim 1 wherein the beverage  
is held under a gaseous pressure in the headspace of at  
least 2.5 atmospheres gauge at 5 to 15° C.  
15
3. A beverage product according to claim 1 or 2, wherein the  
sparingly soluble effervescence inducing gas is selected  
from the group consisting of oxygen, nitrogen, hydrogen,  
noble gases, gaseous hydrocarbons and mixtures thereof.  
20
4. A beverage product according to claim 3, wherein the  
sparingly soluble effervescence inducing gas is oxygen.
5. A beverage product according to any preceding claim, wherein  
25 the beverage has dissolved therein a further gas which is  
more soluble in water than said sparingly soluble  
effervescence inducing gas and which has a partial pressure  
in the headspace not exceeding 1 atmosphere absolute at 18°  
C.  
30
6. A beverage product according to claim 5, wherein the further  
gas is carbon dioxide.
7. A beverage product according to any preceding claim wherein  
35 the valve is an aerosol valve.

8. A beverage product according to any preceding claim, wherein the beverage is water or a tea or coffee based beverage.

5 9. A beverage product according to any preceding claim, wherein the dispenser includes release means which are operable to open the valve to release the liquid from the container, the release means being shaped and positioned for engagement by a user's mouth or teeth to cause or enable release of liquid directly into the user's mouth.

10

10. A beverage product according to claim 8 or 9, wherein the release means includes a button mounted in the nozzle, the button being movable between a valve-closed position and a valve-open position to which it can be moved by a biting action applied across the nozzle.

15

11. A beverage product according to any preceding claim, wherein the container has a dip tube that is attached to the valve and extends into the interior of the container; a closure member normally closing the dip tube and operable from the exterior of the container to open the tube; and an aperture in the tube within the container.

20

12. A beverage product according to claim 1, wherein the gas within the container comprises between 10% and 80% and the drinkable liquid comprises between 20% and 90% of the total volumetric capacity of the container.

25

13. A beverage product according to claim 11 or 12, wherein the quantity of gas expelled from the sealed container is greater than 0.5 cubic centimetres per 1 cubic centimetre of drinkable liquid at atmospheric pressure and 20° C.

30

14. A beverage product according to any one of claims 11 to 13, wherein the aperture is upstream of the closure member.

35

15. A beverage product according to claim 1 that includes means for keeping the valve closed when the dispenser is inverted.
- 5 16. A beverage product according to claim 1 wherein the sparingly soluble effervescence inducing gas is contained in a widget that releases its contents into the container when the valve is opened.
- 10 17. A beverage product according to claim 16 wherein the widget also contains a concentrated flavour that is released into the container when the valve is opened.
- 15 18. A beverage product according to claim 16 wherein the widget contains a substance that adsorbs oxygen.
- 20 19. A method for producing an effervescent beverage fluid comprising holding a beverage within a sparingly soluble effervescence inducing gas dissolved therein in a container of a dispenser with a gaseous pressure in the headspace of the container that is sufficient to cause the beverage to be discharged from the dispenser as an effervescent fluid when the valve is open, and opening that valve to discharge the beverage as said effervescent fluid.
- 25 20. A beverage product substantially as herein described with reference to any one of the accompanying drawings.
- 30 21. A method for producing an effervescent beverage fluid substantially as herein described with reference to any one of the accompanying drawings.

**ABSTRACT**Effervescent beverage product

5

A beverage product that comprises a dispenser and a pressurised beverage. The dispenser has a container for holding the beverage and a valve. The beverage has a sparingly soluble effervescence inducing gas dissolved therein and is held under a gaseous pressure in the head space of the container that is sufficient to cause the beverage to be discharged from the dispenser as an effervescent fluid when the valve is open.

10

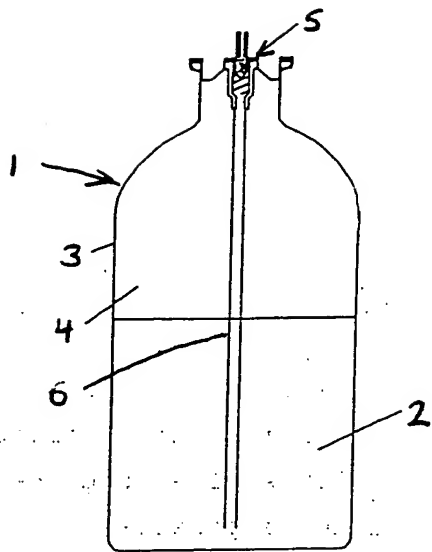


FIG. 1

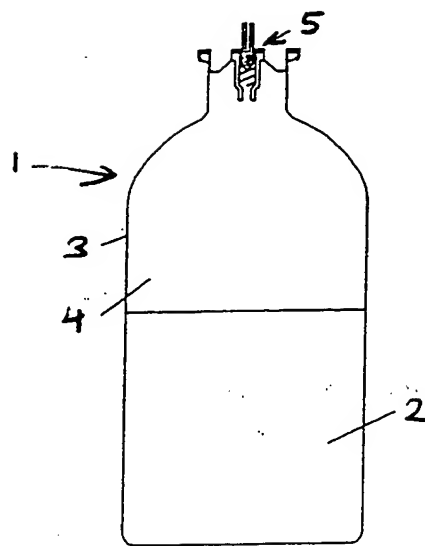


FIG. 3

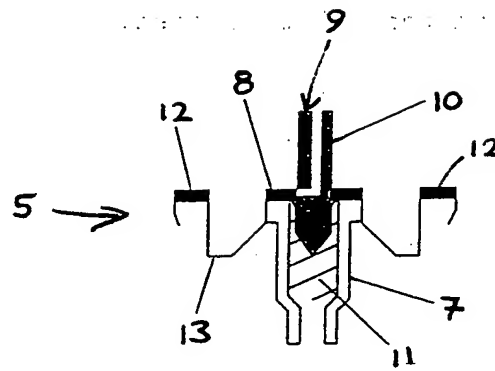


FIG. 2

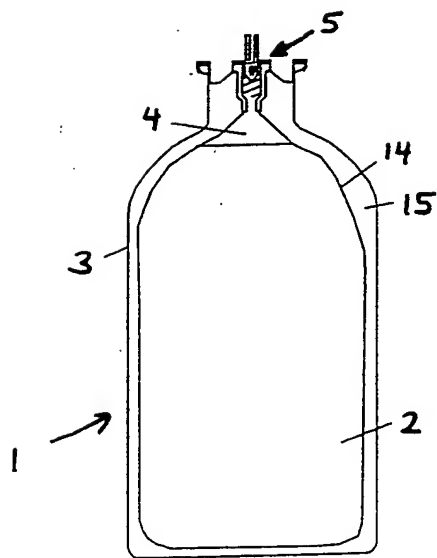


FIG. 4

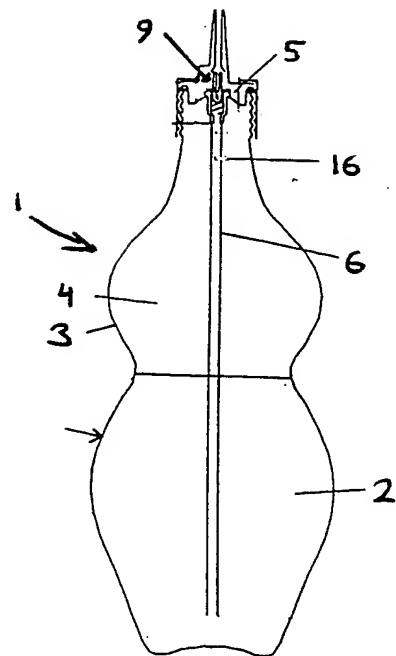
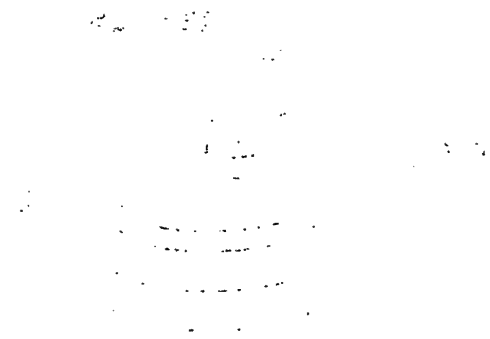
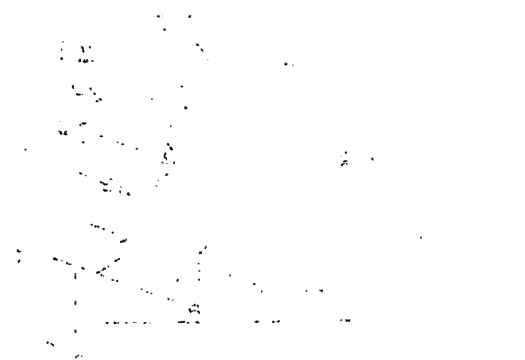
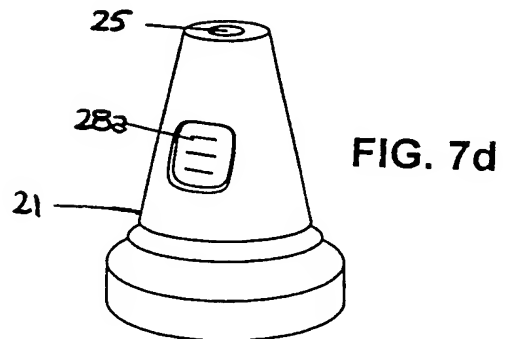
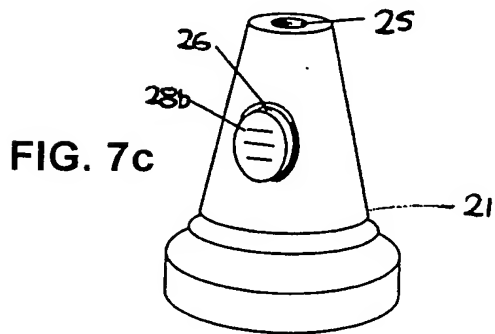
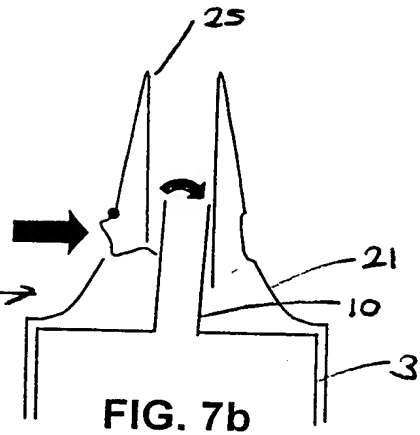
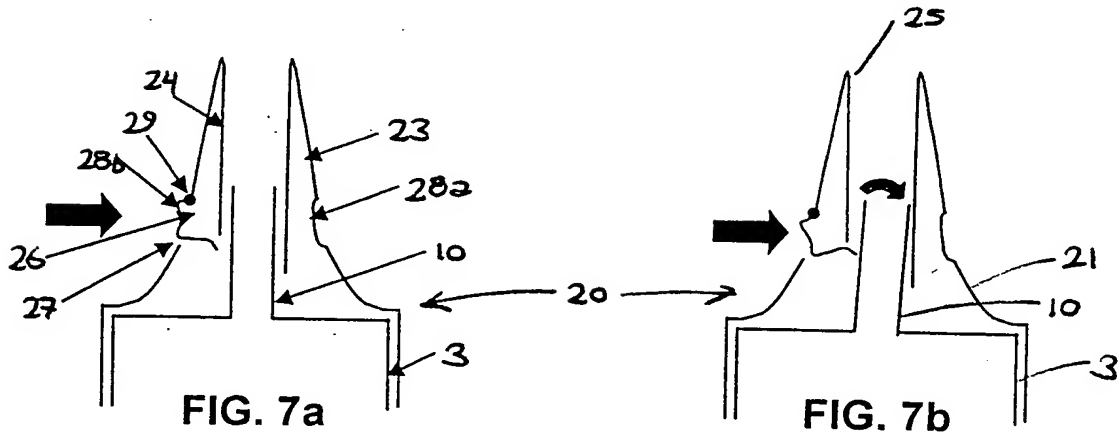
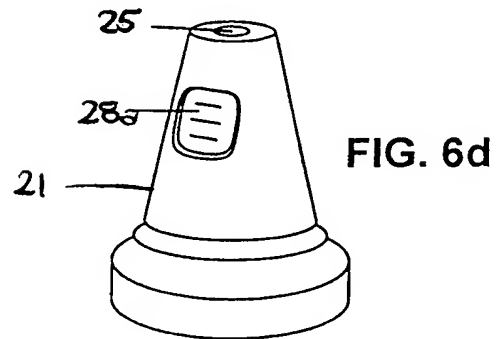
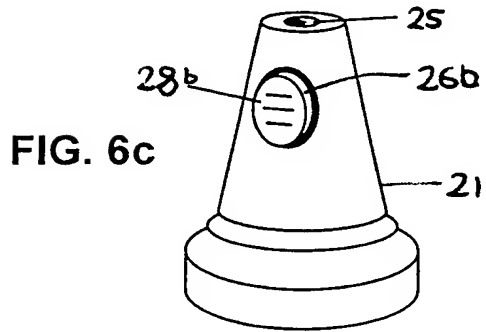
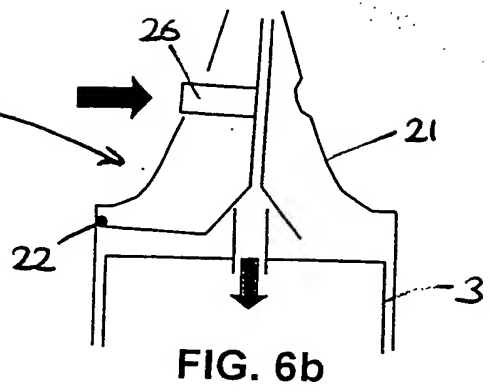
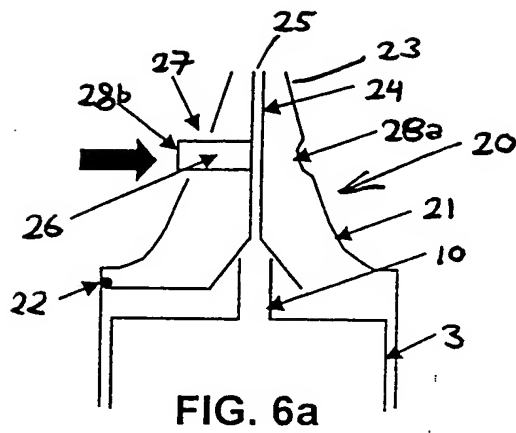


FIG. 5





1. The first part of the report  
describes the general situation  
of the country and the  
main problems which  
are facing it.

2. The second part of the report  
describes the results of the  
survey and the  
main findings of the  
study.

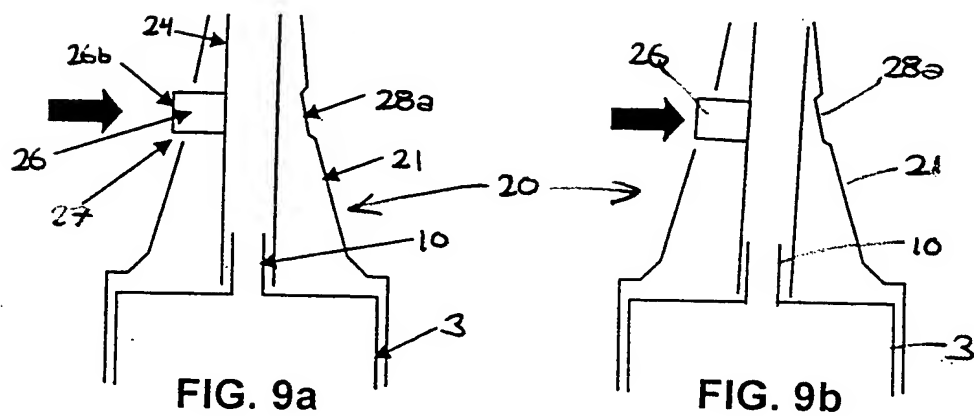
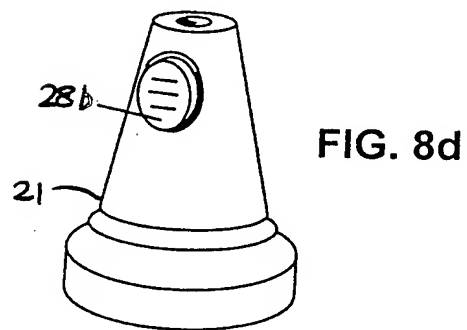
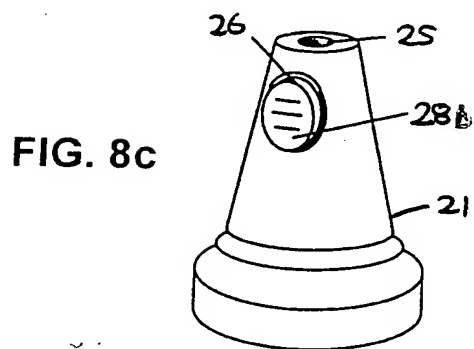
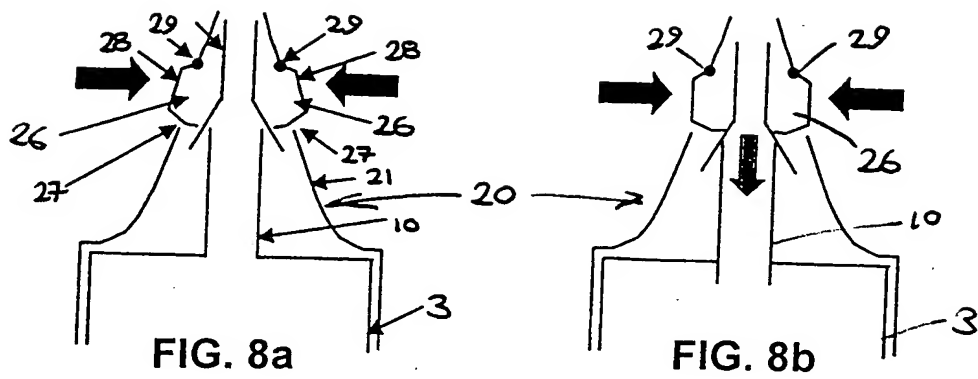
3. The third part of the report  
describes the conclusions  
of the study and the  
main recommendations  
which are being  
made.

4. The fourth part of the report  
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main findings of the  
study and the  
main conclusions  
which are being  
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5. The fifth part of the report  
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6. The sixth part of the report  
describes the  
main findings of the  
study and the  
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which are being  
made.





1. The first part of the document is a list of the names of the persons who were present at the meeting. The names are listed in alphabetical order.

2. The second part of the document is a list of the topics that were discussed at the meeting. The topics are listed in alphabetical order.

3. The third part of the document is a list of the actions that were taken at the meeting. The actions are listed in alphabetical order.

4. The fourth part of the document is a list of the conclusions that were reached at the meeting. The conclusions are listed in alphabetical order.

5. The fifth part of the document is a list of the recommendations that were made at the meeting. The recommendations are listed in alphabetical order.

6. The sixth part of the document is a list of the resolutions that were adopted at the meeting. The resolutions are listed in alphabetical order.

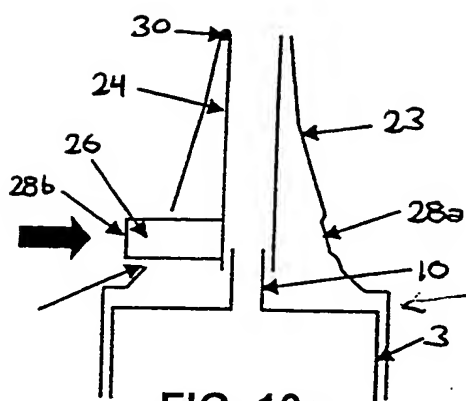


FIG. 10a

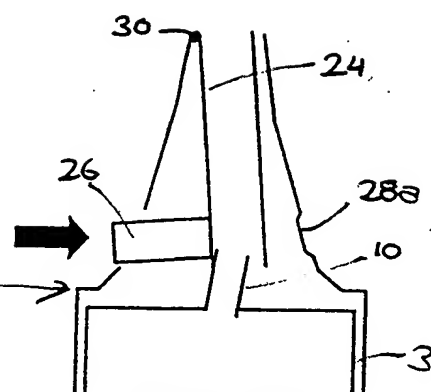


FIG. 10b

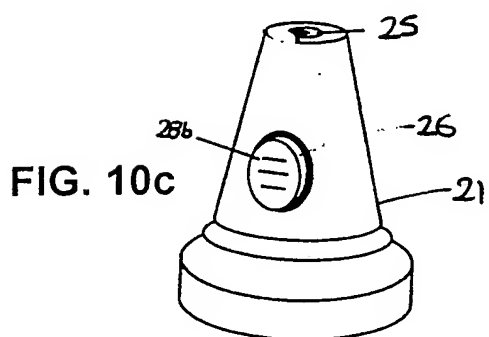


FIG. 10c

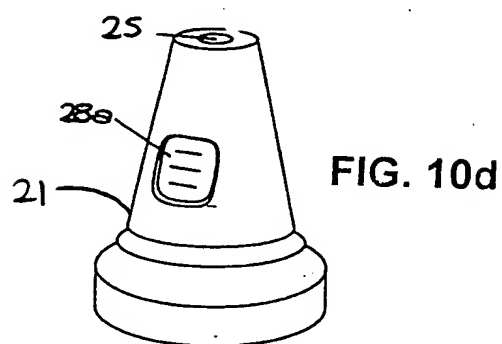


FIG. 10d

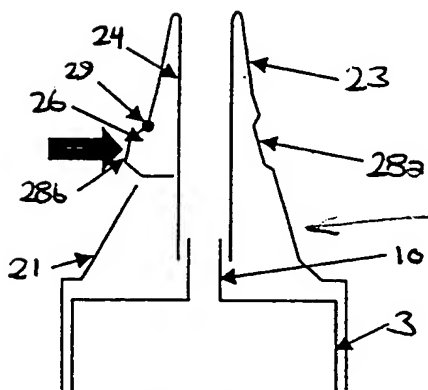


FIG. 11a

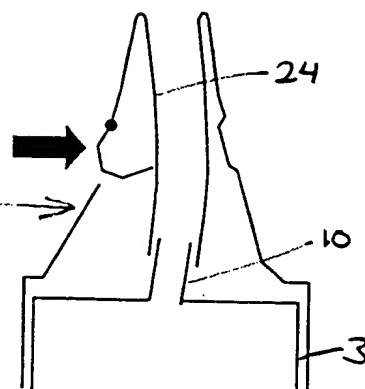
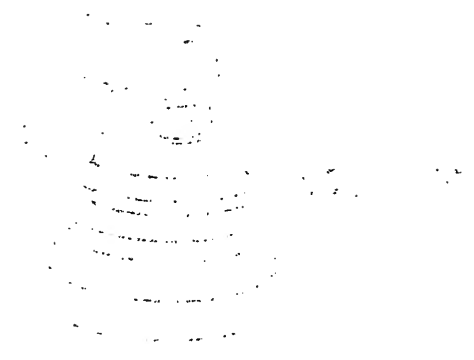
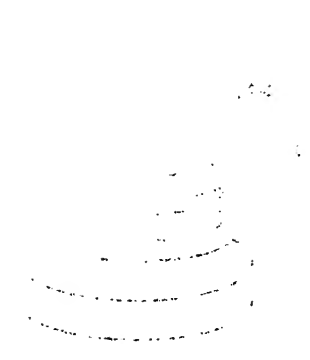
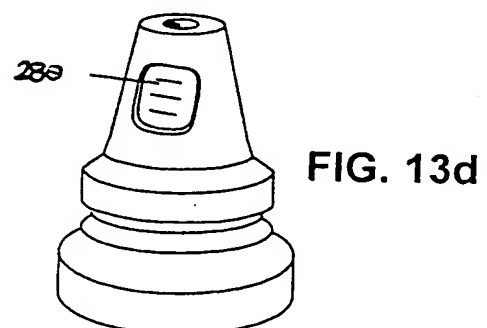
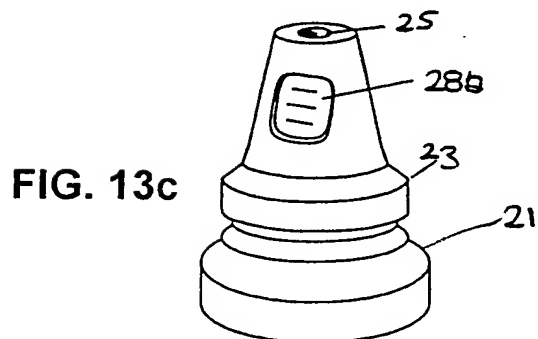
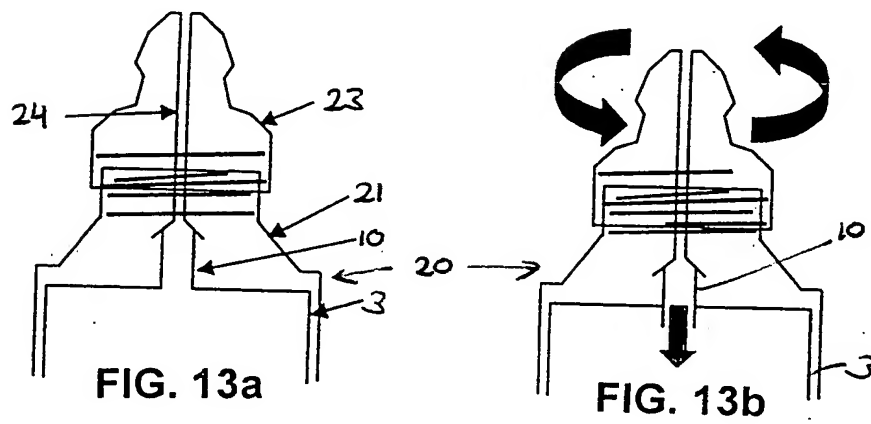
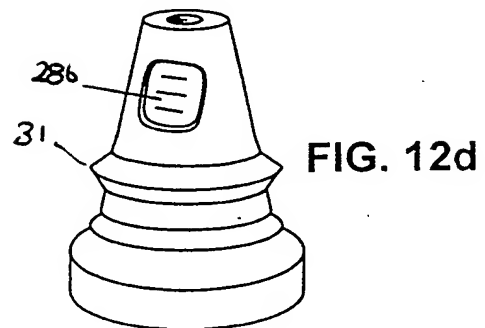
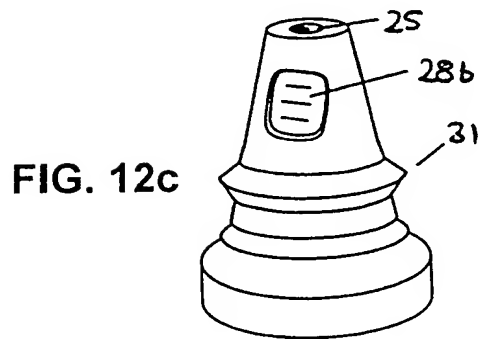
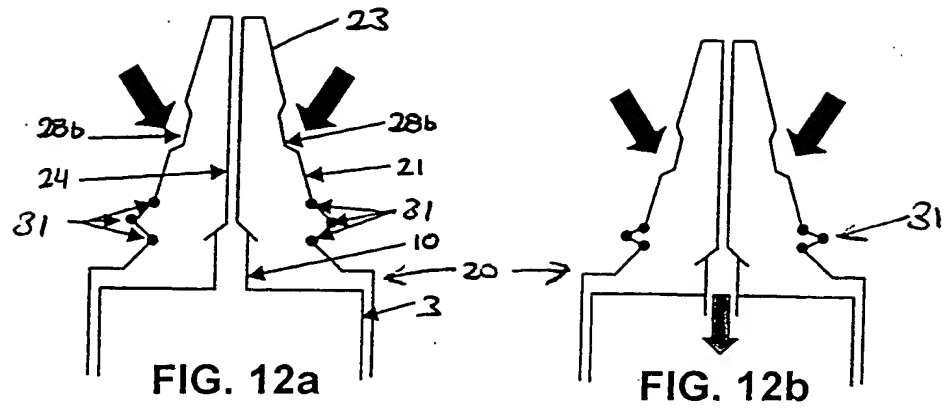


FIG. 11b





1. The first part of the document is a list of names and addresses. The names are: John Doe, Jane Smith, and Bob Johnson. The addresses are: 123 Main St, 456 Elm St, and 789 Oak St.

2. The second part of the document is a list of items and their quantities. The items are: 100 lbs of flour, 50 lbs of sugar, and 25 lbs of butter. The quantities are: 100, 50, and 25.

3. The third part of the document is a list of dates and times. The dates are: 1/1/20, 2/1/20, and 3/1/20. The times are: 10:00 AM, 2:00 PM, and 6:00 PM.

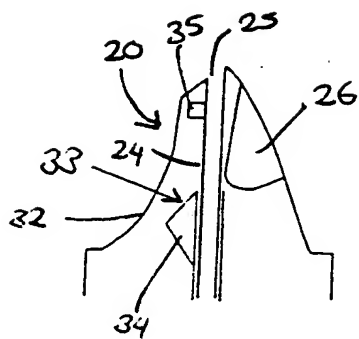


FIG. 14a

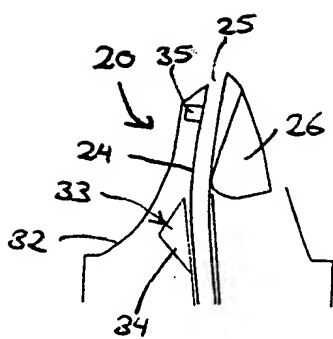


FIG. 14b

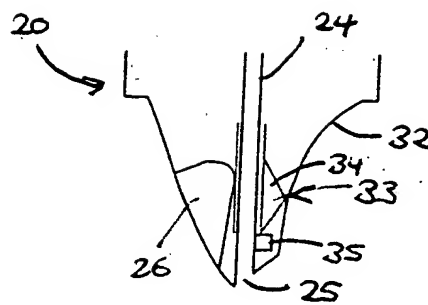


FIG. 14c

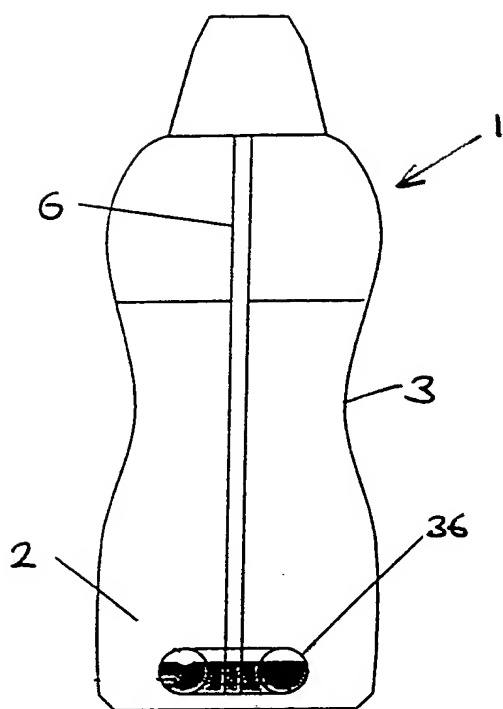


FIG. 15

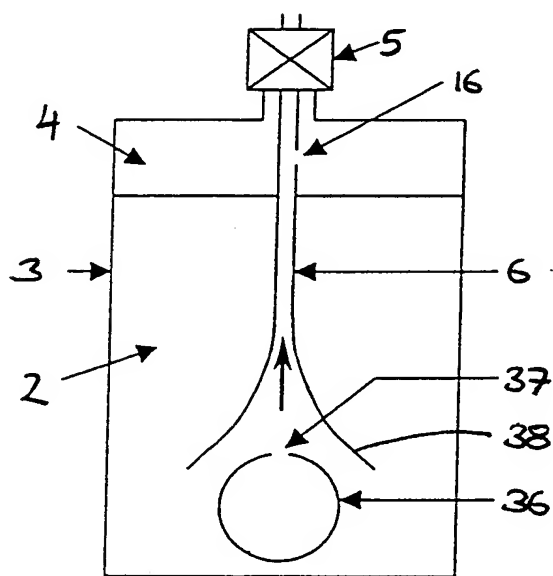


FIG. 17

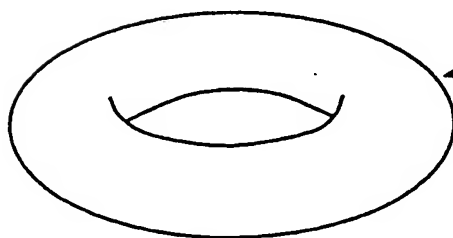


FIG. 16a

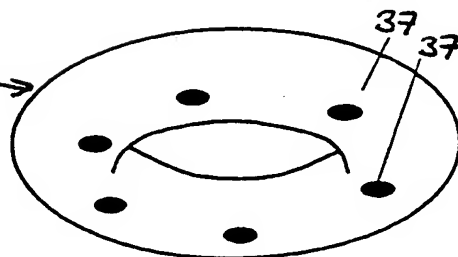


FIG. 16b

1. The first part of the report is a general introduction to the subject of the study. It discusses the importance of the study and the objectives of the research.

2. The second part of the report is a detailed description of the methodology used in the study. It includes information about the sample size, the data collection methods, and the statistical analysis techniques.

1. The first part of the report is a general introduction to the subject of the study. It discusses the importance of the study and the objectives of the research.	2. The second part of the report is a detailed description of the methodology used in the study. It includes information about the sample size, the data collection methods, and the statistical analysis techniques.
3. The third part of the report is a detailed description of the results of the study. It includes information about the mean, standard deviation, and other statistical measures.	4. The fourth part of the report is a detailed description of the conclusions of the study. It includes information about the main findings and the implications of the study.
5. The fifth part of the report is a detailed description of the limitations of the study. It includes information about the sample size, the data collection methods, and the statistical analysis techniques.	6. The sixth part of the report is a detailed description of the recommendations of the study. It includes information about the main findings and the implications of the study.

7. The seventh part of the report is a detailed description of the references used in the study. It includes information about the sources of the data and the literature cited.



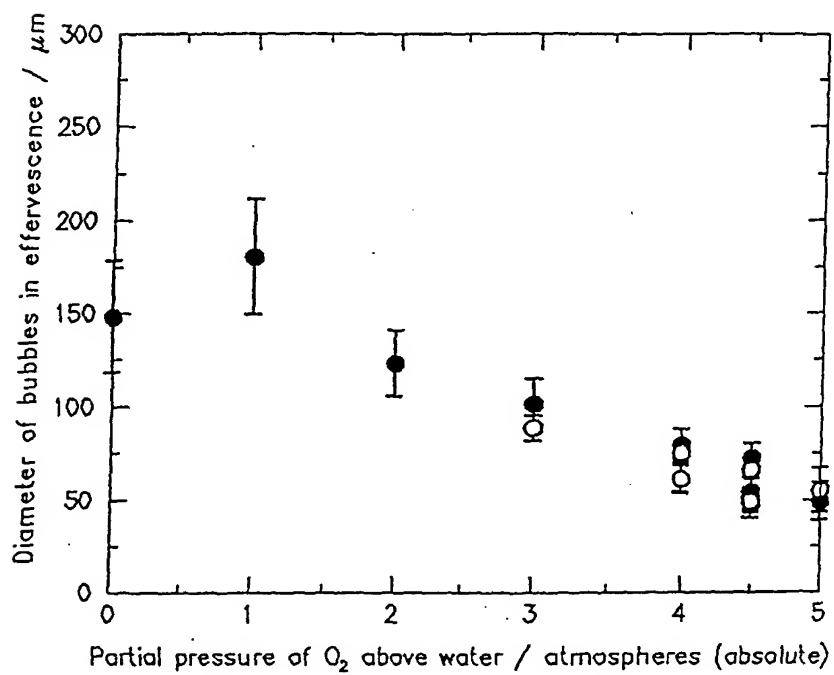
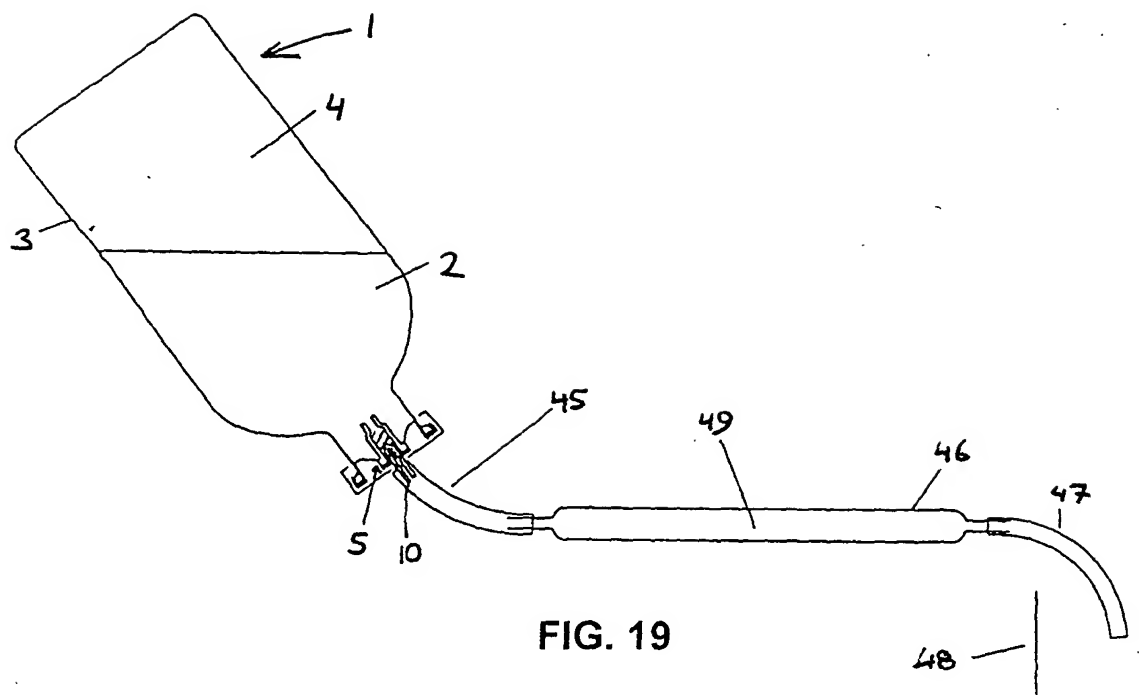
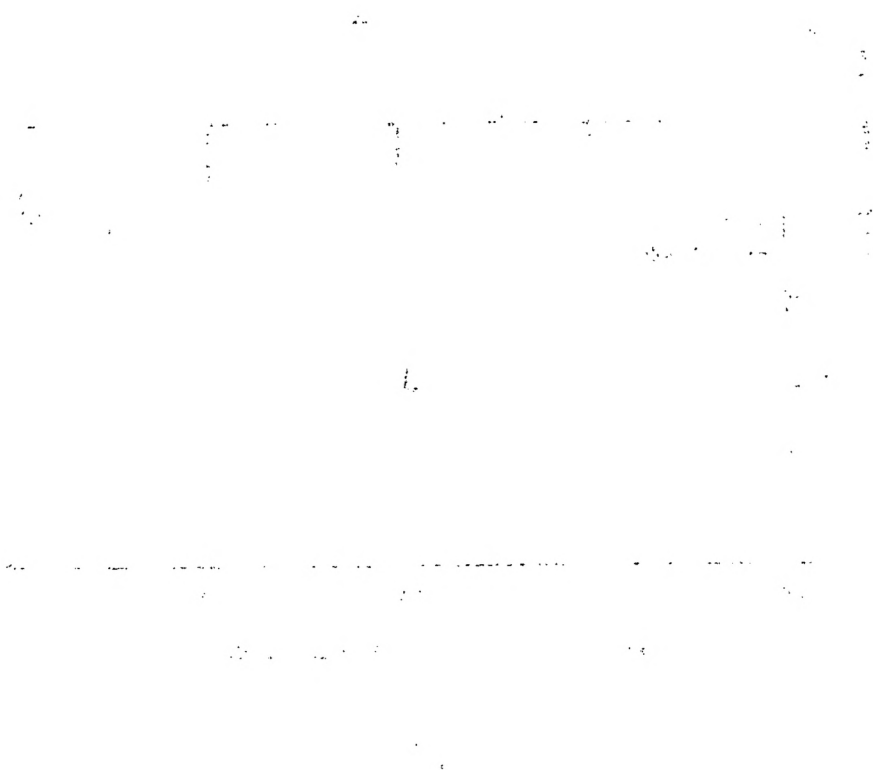


FIG. 20



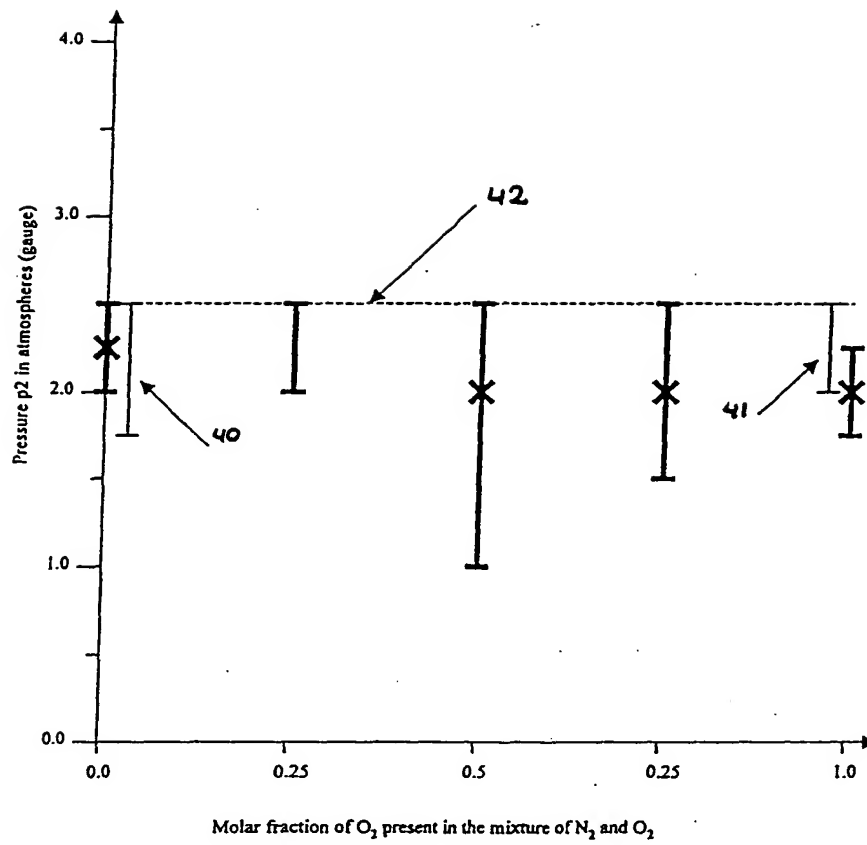


FIG. 18

